

Full Length Article

Contents lists available at [ScienceDirect](www.sciencedirect.com/science/journal/00222313)

Journal of Luminescence

journal homepage: www.elsevier.com/locate/jlumin

Enhanced radioluminescence and improved radioluminescent nuclear battery output performance more than 50% with $SiO₂$ nanosphere coating

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ARTICLE INFO

Keywords: Radioluminescent nuclear battery $SiO₂$ nanosphere coating Electrical output performance enhancement Radioluminescent intensity

ABSTRACT

Radioluminescent nuclear battery as a typical representative of small power device based on the generation and utilization of radioluminescence, which can stable energy supply for equipment in extreme environments. To date, the development of such nuclear battery is primarily limited by the weak radioluminescence intensity. In this work, a method of adding $SiO₂$ nanosphere coating to the fluorescent layer is proposed to solve the problem of weak radioluminescence and significantly improve the battery's output performance. The SiO₂ nanosphere coating can promote the radiative transition rate and quantum efficiency of the ZnS:Cu phosphor of fluorescent layer and the radioluminescent intensity of the ZnS:Cu fluorescent layer increases by 62.32%. And the maximum output power of the corresponding radioluminescent nuclear battery is enhanced by 51.59% compared with that before optimization, under the same X-ray excitation condition.

1. Introduction

As a kind of energy supply device with the advantages of long life and stable output, the nuclear battery can provide energy for space equipment such as satellites and detectors in extreme environments such as deep space [[1](#page-7-0),[2](#page-7-0)]. Radioluminescent nuclear battery (RLNB), as a typical representative of small nuclear battery, shows great potential to provide energy support for aerospace devices in future space exploration [\[3\]](#page-7-0). As shown in [Fig. 1,](#page-1-0) the RLNB can be divided into three main parts: radioactive sources, fluorescent layer, and photovoltaic (PV) modules. The fluorescent layer of the RLNB can convert decay energy from radioactive sources into light energy. Then the PV module converts the light energy into electrical output through the photoelectric effect [[4](#page-7-0)]. And the output performance of the nuclear battery based on indirect energy conversion is much higher than the nuclear battery which converts decay directly into electric energy [[5](#page-7-0)]. The fluorescent layer as the most important part of the RLNB not only protects the PV modules from radiation damage $[6,7]$ $[6,7]$ and increases the service life $[8,9]$ $[8,9]$, but also affects the output performance of the battery. For example, the radioluminescence intensity of the fluorescent layer or the coupling between the response range of the PV module and the spectral range of radioluminescence can determine the electrical output performance of the battery $[10,11]$ $[10,11]$ $[10,11]$. Based on this, it is necessary to take effective measures to enhance the radioluminescence intensity or modulate spectral range of the fluorescent layer, to improve the output performance and expand the application scenarios of the battery $[12-15]$ $[12-15]$.

Many optimization methods for increasing the radioluminescence properties of the fluorescent layer and improving the output performance of the RLNB have been proposed. These methods can be divided into two main categories, one of which is using new fluorescent materials with good properties. Xu et al. $[16,17]$ $[16,17]$ $[16,17]$ proposed a method that prepare a fluorescent layer of perovskite nanocrystals, which can freely adjust emission wavelength. The radioluminescence spectral range of this fluorescent layer matches the response interval of photovoltaic modules better, thus improving the electric output performance of the battery. And Li et al. [\[18](#page-7-0)] doped Mn^{2+} in Cs₃Cu₂I₅ and prepared it as a fluorescent layer to improve the output performance of the RLNB as well as the irradiation stability. Another method is to use existing materials with good luminous properties to improve their radioluminescence intensity through various methods. For example, ZnS:Cu is a fluorescent material that produces yellow-green light with excellent optical properties, energy conversion rate and irradiation resistance [\[19](#page-7-0)–21].

<https://doi.org/10.1016/j.jlumin.2022.119600>

Available online 8 December 2022 0022-2313/© 2022 Elsevier B.V. All rights reserved. Received 22 September 2022; Received in revised form 22 November 2022; Accepted 5 December 2022

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Extensive studies have been carried out for radioluminescent nuclear battery prepared from this fluorescent material [\[22](#page-7-0)–24]. Jiang et al. [[25\]](#page-7-0) proposed a method for irradiation modification of ZnS:Cu phosphor to improve the radioluminescent performance. However, this method is difficult to operate and will change the structure of the material. And Zhang et al. [\[26](#page-7-0)] proposed a method of using Al film as a visible light reflector on ZnS:Cu fluorescent layer. This method not only does not change the structure of the material itself but also increases the maximum output of the battery by 34.22%. On the basis of these enhancement methods, we propose a method to enhance the radioluminescence properties of the ZnS:Cu fluorescent material without changing its properties and structures.

Specifically, we add $SiO₂$ nanosphere coating to the fluorescent layer by gravitational deposition method and spin-coating process. This method successfully enhances the radioluminescence performance of ZnS:Cu fluorescent layer by enhancing the quantum efficiency of fluorescent materials and improves the electrical output performance of the RLNB without changing the properties and structure of the material. And the structure of the fluorescent layer in this work has a much more significant improvement in the RLNB output performance than the current fluorescent layer structure.

2. Materials and method

2.1. Preparation of fluorescent layer with SiO2 nanosphere coating

Approximately 3.5 mL of ammonia (25%, AR) and 1.5 mL of deionized water are mixed and heated to 303 K and 25 mL of anhydrous ethanol (99.7%, AR) and 2.4 mL of tetraethyl orthosilicate (99%, GC) are mixed and sonicate for 30 min, then add to the deionized water mixture and stir at 303 K for 12 h. The precipitate is obtained by centrifugal separation and drying. The $SiO₂$ powders are ultrasonically dispersed in anhydrous ethanol at 2 wt%. The glass substrate is placed vertically into the solution, then place in a blast drying oven and dry for 3 h for preparation of the $SiO₂$ nanosphere coating.

Polystyrene and xylene (99%, AR) are mixed in the ratio of 1 g:2.5 mL and stirred for 12 h for the preparation of a transparent liquid. The ZnS:Cu phosphors (Purchased from Guangzhou Chuang Rong Chemical Technology Co, particle size of $15 \mu m$) and the prepared solution are mixed according to the mass ratio of 1:1 and use the magnetic mixer to mix it fully for 30 min. The mixture dropwise on the $SiO₂$ nanosphere coating and spin-coating for 5 s at a speed of 1000 rpm to cover the surface of the coating. Then the fluorescent layer is cured by spincoating at 5000 rpm for 90 s. And the fluorescent layers without SiO2 nanosphere coating are prepared by dropping the mixture directly onto the glass substrate and using the same spin-coating process.

2.2. FDTD simulations of fluorescence enhancement

The simulation is performed using the Finite Difference Time

Domain (FDTD) method. The emission process of ZnS:Cu is simulated using an electric dipole set at the coordinates of (0, 0, 0) μm with a wavelength range of 400–650 nm and the emission direction along the z axis. At a distance of 15 nm from the dipole, random distribution of $SiO₂$ nanospheres with a diameter of 450 nm is set up, where the center coordinate of the centermost $SiO₂$ nanosphere is $(0, 0, -0.24)$ µm. And a perfectly matched layer (PML) is used at the x, y and z boundaries of the SiO2 nanospheres to absorb the light. An electric field monitor is placed at $z = 0$ µm to analyze the electric field distribution of the fluorescence emission process. A power monitor analysis group with center coordinates at $(0, 0, -0.2)$ μm and span of 0.6 μm is set up to obtain the output power when the output electric dipole interacts with SiO₂. A smaller power monitor analysis group set with span of 0.015 μm and centered at (0, 0, 0) μm is set to obtain the output power of the electric dipole alone.

The change of quantum efficiency η after adding $SiO₂$ nanospheres is calculated using the method described in references [27–[30\]](#page-7-0).

$$
\eta = \frac{\Gamma_{\text{rad}}}{\left(\Gamma_{\text{tot}} + \left(1 - \eta_{\text{i}}\right) / \eta_{\text{i}}\right)}\tag{1}
$$

$$
\Gamma_{\rm rad} = P_{\rm rad}/P_0 \tag{2}
$$

$$
\Gamma_{\text{tot}} = P_{\text{tot}} / P_0 \tag{3}
$$

where Γ_{rad} represents radiative decay rate, Γ_{tot} represents total decay rate, and η_i represents initial quantum efficiency (which is set to 13.3% in this work [[31\]](#page-7-0)).

2.3. Optical and electrical properties testing

An electron-multiplying charge-coupled devices (ECCD) camera (Andor iXon Ultra 888, USA) equipped with a Canon EF 24–70 mm f/ 2.8L II USM zoom lens is used to record the images of the fluorescent layer. This work uses the X-ray to excite fluorescent materials to produce radioluminescence, and the voltage and current of the X-ray tube mainly control its intensity. The specific test environment as shown in [Fig. 2](#page-2-0). The ZnS:Cu fluorescence layer is located directly in front of the X-ray tube gate, and the lens of the ECCD camera is pointed at the fluorescence layer. When the ZnS:Cu material is excited by the X-ray, the radioluminescent photons emitted from the surface of the fluorescence layer are captured by the ECCD camera. And the radioluminescence spectra is obtained by Agilent Technologies Cary Eclipse fluorescence spectrophotometer. The images taken by the ECCD camera can be processed by ImageJ software to obtain the intensity of the radioluminescence photons generated on the surface of different fluorescent layers.

Based on the previous work about RLNB preparation and performance optimization, AlGaInP-based PV modules are chosen in this work. And the AlGaInP-based PV modules show good photovoltaic conversion performance and have a good matching with radioluminescence spectroscopy of ZnS:Cu. On this basis, we explore the enhancement

Fig. 1. Working mechanism of the radioluminescent nuclear battery.

Fig. 2. Testing environment of photons intensity on the surface of fluorescent layer.

mechanism of fluorescent layer, and compare its effect with different optimization methods. The electrical property of the RLNB is evaluated by the Keithley 2636A source meter.

3. Results and discussion

3.1. Structural characterization

Fig. 3 shows the SEM images of the ZnS: Cu fluorescent layer and $SiO₂$ nanosphere coating. From the image of the cross-section of the prepared ZnS:Cu fluorescent layer, the structure of fluorescent layer is divided into three parts: glass substrate, $SiO₂$ nanosphere coating and ZnS:Cu phosphor layer. The ZnS:Cu phosphor layer and $SiO₂$ nanosphere coating are just closely adhered to each other, and do not change the structure of the ZnS:Cu phosphor. The thickness of the $SiO₂$ nanosphere coating is about 80 μ m and the SiO₂ nanospheres in the coating are randomly distributed. In addition, the glass substrate only plays a supporting role so other materials can replace it according to the actual situation.

[Fig. 4a](#page-3-0) shows the average particle size of the $SiO₂$ nanosphere particles is 450 nm in the coating. The photoluminescence (PL) spectrum of the ZnS:Cu (excited by the 300 nm UV) and the reflection spectrum of $SiO₂$ nanosphere coating prepared in this work, as shown in [Fig. 4b](#page-3-0). The ZnS:Cu has one emission peak centered at 518 nm and these two spectra have a large overlap from 400 to 600 nm, which is beneficial for the interaction of the ZnS:Cu and $SiO₂$ nanospheres coating.

3.2. Radioluminescence analysis

The radioluminescence properties of the fluorescent layers prepared in this work under different X-ray excitation conditions are further analyzed. [Fig. 5](#page-3-0) represents the ECCD camera recorded images of radioluminescent photons emitted from the surface of fluorescent layers under the X-ray with different voltages of X-ray tube (40 kV, 50 kV, 60 kV) and the current of X-ray tube is kept at 0.8 mA. And the $SiO₂$ nanosphere coating is added to the fluorescent layer in [Fig. 5](#page-3-0)b. When the ZnS:Cu fluorescent layer is not excited, almost no radioluminescent photons are produced. However, when it is excited by the X-ray, a large number of radioluminescent photons are produced on the surface. The number of photons from the fluorescence layer with $SiO₂$ nanosphere coating is more than the fluorescence layer without $SiO₂$ nanosphere coating in the same situation. They all increase with the increase of the voltage of the X-ray tube.

[Fig. 6](#page-3-0)a shows the variation of the radioluminescent photon intensity produced by fluorescent layers varied with the X-ray intensity. It can be seen that the radioluminescent photon intensity produced by the fluorescent layer increases with the X-ray intensity. And under the same excitation conditions, the photon intensity of ZnS:Cu fluorescent layer with $SiO₂$ nanosphere coating is greater than fluorescent layer without coating. The radioluminescence spectra of the fluorescent layer excited by X-ray is measured when the voltage of X-ray tube is 60 kV and the current of X-ray tube is 0.8 mA, as shown in [Fig. 6](#page-3-0)b. Under the excitation of X-rays, the emission peak of ZnS:Cu is concentrated at 526 nm. The SiO2 nanosphere coating does not emit fluorescence under X-ray excitation and does not affect the peak position of the ZnS:Cu radioluminescence spectrum. Thus, adding it to the fluorescent layer serves to increase the radioluminescence intensity and does not significantly affect the luminous properties of the ZnS:Cu fluorescent material itself. Moreover, when the voltage of the X-ray tube reaches 60 kV, the radioluminescence intensity of the fluorescent layer increases by 62.32% with the addition of $SiO₂$ nanosphere coating.

3.3. Enhancement mechanism analysis of radioluminescence

The model is established in FDTD based on the experimental results, as shown in [Fig. 7](#page-4-0). It contains randomly distributed $SiO₂$ nanospheres and a dipole is located directly above them. The luminescence phenomena and process of radioluminescence and photoluminescence are similar and both generate electron-hole pairs. Electron-hole pair composite and de-excitation produce radioluminescence [[32\]](#page-7-0). Therefore, dipoles are used instead of electron-hole pairs in this work to simulate the radioluminescence process [\[33](#page-7-0)].

As shown in [Fig. 8](#page-4-0)a, the electric field intensity gradually decreases from the inside to the outside and uniformly distributed, for the case of only dipole. [Fig. 8](#page-4-0)b describes the field intensity of dipole with $SiO₂$ nanospheres. The fluorescence resonance between $SiO₂$ nanospheres and dipoles changes the electric field distribution. So the distribution range of the electric field becomes larger, from regular circular to

Fig. 3. SEM characterization images of the cross-section of (a) the prepared ZnS:Cu fluorescent layer and (b) the surface of the prepared SiO₂ nanosphere coating.

Fig. 4. (a) Particle size distribution of SiO₂ nanospheres. (b) The reflection spectrum of SiO₂ nanosphere coating and normalized PL spectrum of the ZnS:Cu.

Fig. 5. Radioluminescent images of ZnS:Cu fluorescent layers (a) without SiO₂ nanosphere coating and (b) with SiO₂ nanosphere coating under different X-ray excitation conditions.

Fig. 6. (a) The radioluminescent photon intensity of fluorescent layers under different radiation conditions. (b) RL spectra of two kinds of fluorescent layers and the SiO2 nanosphere coating.

Fig. 7. (a) Model for FDTD simulation and (b) the details of the SiO₂ nanospheres model in 3D space.

Fig. 8. The electric field intensity distribution of (a) a dipole without SiO₂ and (b) a dipole with SiO₂.

irregular and the electric field intensity is significantly higher at the same position than without adding $SiO₂$ nanospheres. The change of electric field intensity also affects the excitation rate. When the

Fig. 9. The influence of different fluorescence emission directions on the enhancement of quantum efficiency and the inset shows the electric field distribution when dipoles emit fluorescence along different directions.

excitation rate increases, it will affect the spontaneous decay rate and enhance the fluorescence intensity [\[34](#page-7-0)–36].

The radioluminescence produced by the phosphor is isotropic and diverges around. Therefore, the change of quantum efficiency and electric field distribution are analyzed when the dipole is emitted along the three directions of the x, y and z axis, as shown in Fig. 9. It can be seen that when the dipole emits fluorescence along the z axis direction, its electric field intensity is much greater than along the x and y axis directions. The reason for this phenomenon is that the fluorescence produced by the dipole interacts with the $SiO₂$ nanospheres to produce resonance, which changes the field intensity, and improves the quantum efficiency [[37](#page-8-0)–39]. According to the simulation results, when the fluorescence wavelength is 526 nm, the quantum efficiency of the emitter along the z axis increases by 79.2%. But when it emits along the x and y axis, the quantum efficiency only increases by 11.4% and 12.1%. Therefore, when the fluorescence emission direction is perpendicular to the $SiO₂$ nanospheres, the resonance phenomenon is most obvious, which also corresponds to the electric field distribution. Based on the above study, the method of adding $SiO₂$ nanosphere coating to the fluorescent layer is an effective way to enhance the radioluminescence intensity without changing the inherent properties of fluorescent material.

3.4. Application of SiO2 nanosphere coating in radioluminescent nuclear battery

Based on the above-mentioned optical performance study, the electrical performance testing of nuclear batteries through different energy conversion mechanisms is carried out. According to the different energy conversion mechanisms, the nuclear batteries can be divided into three types. 1) A nuclear battery without a fluorescent layer directly converts X-ray energy into electrical energy. 2) A nuclear battery with a fluorescent layer indirectly converts the energy of X-rays into electrical energy. 3) A nuclear battery decorated with a $SiO₂$ nanosphere coating on the fluorescent layer indirectly converts X-ray energy into electrical energy. As shown in Fig. 10a–e, the electrical output performance of RLNBs is tested when the voltage of X-ray tube at 40–60 kV, and the tube current of X-ray is kept at 0.8 mA. It can be seen that the electrical output performance of the nuclear battery with fluorescent layer is much greater than that without fluorescent layer. It is because of the strong penetration of X-ray, which can deposit energy in both the fluorescent layer and the PV module, thus making the electrical output performance of the battery stronger. And the electrical output performance of nuclear batteries increases with the X-ray intensity, regardless of the type of energy conversion based.

Fig. 10f shows the variation in output power density among the three kinds of nuclear batteries at different voltages of X-ray tube changes and fits the variation in output power density. It can be seen that adding fluorescent layer can be very effective in improving the output power density of the nuclear battery. The output power density of the nuclear batteries with the additional fluorescent layer can be increased from 0.91 nW/cm² to 2.84 nW/cm², a 3.12-fold enhancement, when the voltage of X-ray tube is 60 kV. And the output power density of nuclear battery increased to 4.28 nW/cm² after using fluorescent layer with $SiO₂$

Fig. 10. *I*–*V* and *P*–*V* curves of the electrical output of radioluminescent nuclear battery with the current of X-ray tube is 0.8 mA and the voltage of tube is (a) 40 kV, (b) 45 kV, (c) 50 kV, (d) 55 kV, (e) 60 kV. (f) Variation of the output power density of the battery at different voltages of tube.

nanosphere coating, showing 4.7-fold enhancement, due to the enhancement of radioluminescence by the $SiO₂$ nanospheres coating. In addition, it can be derived from the trend of output power density that with the further increase of X-ray intensity, the difference of output power density between nuclear batteries with fluorescent layer and nuclear battery without fluorescent layer will be greater. The addition of SiO2 nanosphere coating does not affect the change of this trend and further increases this difference. Therefore, using $SiO₂$ nanosphere coating to improve the radioluminescence intensity is beneficial to improve the output performance of RLNB.

3.5. Comparing the effect of different fluorescent layers optimization schemes on the batteries

The effects of different fluorescent layer structures on the electrical output performance of RLNBs are compared. As shown in Fig. 11a, four kinds of fluorescent layers are prepared for the comparison in terms of enhancement in RLNB electrical output performance. #1 is a ZnS:Cu fluorescent layer with glass as the substrate, #2 is a ZnS:Cu fluorescent layer with Al as the substrate, #3 is a ZnS:Cu fluorescent layer with glass as the substrate and $SiO₂$ nanosphere coating added, and #4 is an Al film added behind the fluorescent layer of #1.

The results of the test on electrical output performance are shown in Fig. 11b–e. The electrical output performance of the RLNB can be effectively enhanced by adding Al film or $SiO₂$ nanosphere coating. The electrical output performance gradually increases with the voltage of Xray tube in the four kinds fluorescent layer structures. To obtain a good reflection performance of visible light for the Al film used in RLNB #4, we set the thickness at 0.15 mm. The thickness of Al film is thicker than that of $SiO₂$ nanosphere coating, and it completely covers the fluorescent layer. In RLNB #2, the thickness of Al layer is 2 mm to provide good support for the ZnS:Cu layer. And the Al layer used in RLNB #2 is much thicker than that of RLNB #4, but the electrical output performance of RLNB $#4$ is higher than that of RLNB $#2$. It is because the ZnS:Cu phosphor in the fluorescence layer of RLNB #2 is in direct contact with Al and makes a decrease in the radioluminescence intensity produced by

the fluorescence layer, due to the phenomenon of fluorescence quenching appearances [[40\]](#page-8-0). And from the test results, under the same radiation conditions, the $I_{\rm sc}$ value of the RLNB with SiO_2 nanosphere coating is higher than that of the RLNB with Al film as the reflector, while the V_{oc} is almost the same probably because the $SiO₂$ nanosphere coating induces the fluorescence layer to produce a higher radioluminescence intensity. Table 1 lists the *I*sc and *P*max values of four kinds of RLNBs when the voltage of X-ray tube is 60 kV and current of X-ray tube is 0.8 mA. It can be seen that the P_{max} value and I_{sc} value increased by 35.68% and 23.32% due to the structure of using Al film. However, under the same condition, the fluorescent layer structure with the addition of $SiO₂$ nanosphere coating results in a greater increase in RLNB output performance, the P_{max} value increase of up to 51.59% and the I_{sc} value to 35.63%. The results show that the design and optimization of the fluorescence layer structures can indeed enhance the electrical output performance and the $SiO₂$ nanosphere coating can also enhance the radioluminescence intensity without changing the other properties of the material.

4. Conclusion

In this work, we propose a method for enhancing the radioluminescence intensity without changing the structure and properties of the ZnS:Cu phosphor. The specific approach is to gravitational deposition method the $SiO₂$ nanosphere coating on the substrate, and then prepare the ZnS:Cu fluorescent layer by spin-coating method. The increase in the radioluminescence intensity of the fluorescent layer by 62.32% is attributed to the enhancement of the quantum efficiency of

Table 1

The I_{sc} and P_{max} value of the radioluminescent nuclear battery at the voltage of X-ray tube is 60 kV and the current of X-ray tube is 0.8 mA.

Sample	RLNB #1	RIMB#2	RLNB #3	RLNB #4
I_{sc} (nA)	10.72	12.77	14.54	13.22
P_{max} (nW)	2.83	3.67	4.29	3.84

Fig. 11. (a) Four different structures of ZnS:Cu fluorescent layers. Electrical properties of the radioluminescent nuclear battery with different fluorescent layers under different voltage of the X-ray tube, (b) short circuit current *I_{so}* (c) open circuit voltage *V*_{oc}, (d) maximum output power *P*_{max}, and (e) fill factor *FF*.

the ZnS:Cu phosphor by the $SiO₂$ nanospheres. And the method can increase the maximum output power of the radioluminescent nuclear battery by 51.59%, when X-ray is used as the source of radiation. And the results from a series of test data indicate that the radioluminescent nuclear battery's output performance increases as the X-ray intensity increases and the $SiO₂$ nanosphere coating can further improve the performance amplification. Moreover, the electrical output performance of the radioluminescent nuclear battery can also be improved by changing the structure of the fluorescent layer. This method of our work has more advantages compared with other fluorescent layer structure design methods. The method of adding $SiO₂$ nanosphere coating to enhance the radioluminescence of ZnS:Cu is not only of great significance to increasing electrical performance output of the radioluminescent nuclear battery, but also helpful to the applications in other areas, such as lighting devices, optoelectronic devices and detectors.

Credit author statement

Yuhang He: Conceptualization, Methodology, Formal analysis, Investigation, Writing-Original draft, Writing–Review & Editing; **Zhiheng Xu:** Validation, Data curation, Formal analysis, Writing-Reviewing and Editing, Funding acquisition, Supervision; **Hongyu Wang:** Visualization; **Mingxin Bian:** Investigation; **Yunpeng Liu:** Funding acquisition, Supervision; **Xiaobin Tang:** Writing-Reviewing and Editing, Supervision, Project administration.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Data availability

Data will be made available on request.

Acknowledgements

This work is supported by the National Natural Science Foundation of China (Grant No. 12005101), the Natural Science Foundation of Jiangsu Province (Grant No. BK20201288), China Postdoctoral Science Foundation (Grant No. 2022M711613) and the Excellent Postdoctoral Program of Jiangsu Province (Grant No. 2022ZB235).

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